



Univerza v Mariboru

Fakulteta za naravoslovje
in matematiko

UČNI NAČRT PREDMETA / COURSE SYLLABUS

Predmet: Kemija

Course title: Chemistry

Študijski program in stopnja Study programme and level	Študijska smer Study field	Letnik Academic year	Semester
Enovit magistrski študijski program druge stopnje Predmetni učitelj	/	1.	1.
Five-year master's degree program Subject Teacher	/		

Vrsta predmeta / Course type

Obvezni / Obligatory

Univerzitetna koda predmeta / University course code:

Predavanja Lectures	Seminar Seminar	Vaje Tutorial	Lab. vaje Laboratory work	Terenske vaje Field work	Samost. delo Individ. work	ECTS
45			30		105	6

Nosilec predmeta / Lecturer:

Matjaž KRISTL

Jeziki / Predavanja / Lectures: slovenski / slovene

Languages: Vaje / Tutorial: slovenski / slovene

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Jih ni.

Prerequisites:

No.

Vsebina:

V prvem delu so obravnavane osnovne zakonitosti in zakoni splošne kemije:

- Lastnosti snovi, kemijske formule in enačba, osnove stehiometrije
- Plinski zakoni, idealno in neidealno obnašanje plinov
- Struktura atomov, periodni sistem elementov
- Kemijske vezi in njihove lastnosti
- Rastopine, solvatacija, koncentracije raztopin, topnost
- Kisline, baze, pH, pufri

Poudarek drugega dela je na osnovah anorganske kemije, vključno z osnovami anorganske nomenklature:

- Vodik, kisik, voda
- Elementi VII. skupine: F, Cl, Br, I
- Elementi VI. skupine: S
- Elementi V. skupine: N, P
- Elementi IV. Skupine: C, Pb
- Elementi I.– III. skupine: Na, K, Ca, Mg, Al

Tretji del je namenjen osnovam organske kemije in nomenklature:

- Alkani, alkeni, alkini, izomerija
- Aromatske spojine
- Organske kisline
- Alkoholi, ogljikovi hidrati
- Etri, estri, lipidi
- Aminokisline, peptidi, proteini

Content (Syllabus outline):

In the first part, basic principles and laws of general chemistry are discussed:

- Properties of matter, chemical formulas and equations, basics of stoichiometry
- Gas laws, ideal and non-ideal behaviour of gasses
- Structure of atoms, periodic table of elements
- Chemical bonds and their properties
- Solutions, solvation, concentration of solutions, solubility
- Acids, bases, pH, buffer solutions

The emphasis of the second part is on the basics of inorganic chemistry, including basic inorganic nomenclature:

- Hydrogen, oxygen, water
- Group VII elements: F, Cl, Br, I
- Group VI elements: S
- Group V elements: N, P
- Group IV elements: C, Pb
- Group I – III elements: Na, K, Ca, Mg, Al

The third part is devoted to basic organic chemistry and organic nomenclature:

- Alkanes, alkenes, alkynes, isomerism
- Aromatic compounds
- Organic acids
- Alcohols, carbohydrates
- Ethers, esters, lipids
- Amino acids, peptides, proteins

Temeljni literatura in viri / Readings:

- F. Lazarini, J. Brenčič, Splošna in anorganska kemija, DZS Ljubljana (1992)
- P.W. Atkins, Kemija - zakonitosti in uporaba, TZS Ljubljana (1995)
- S. Pine, Organic Chemistry 5th edition, McGraw Hill (1996)

Cilji in kompetence:

- Kandidat bo seznanjen z osnovnimi pojmi splošne, anorganske in organske kemije
- Kandidat bo razumel vplive strukture spojin na njihove lastnosti in reaktivnost
- Kandidat bo spoznal kemizem najpomembnejših elementov in spojin prisotnih v vsakdanjem življenju, v okolju, laboratoriju in kemijski industriji

Objectives and competences:

- The candidate will get acquainted with the basic conception of general, inorganic and organic chemistry
- The candidate will understand the influences of the structure of compounds on their properties and reactivity
- The candidate will know the chemistry of the most important elements and

compounds in the environment, chemical laboratory and industry

Predvideni študijski rezultati:

Intended learning outcomes:

Znanje in razumevanje:

Poznavanje osnovnih principov splošne, anorganske in organske kemije. Poznavanje pridobivanja, lastnosti in glavnih spojin izbranih reprezentativnih elementov, s poudarkom na spojinah z vodikom in kisikom. Obvladovanje osnovnega kemijskega računanja na osnovi kemijskih enačb. Razumevanje periodičnih lastnosti elementov po periodnem sistemu. Razumevanja okoljskih ciklov nekaterih elementov.

Prenosljive/ključne spretnosti in drugi atributi:

Pridobitev osnovnega kemijskega znanja potrebna za sodelovanje pri predmetih ki zahtevajo kemijsko predznanje. Prepoznavanje elementov in spojin v okolju in njihovega vpliva. Poznavanje in predvidevanje vpliva sintetičnih spojin na okolje.

Knowledge and understanding:

Knowledge about basic principles of general, inorganic and organic chemistry. Production, properties and main compounds of selected main group elements, with the focus on hydrogen and oxygen compounds. Performing basic calculations based on chemical equations. Understanding of periodicity of element properties throughout the periodic table. Understanding of environmental elemental cycles.

Transferable/Key Skills and other attributes:

Acquirement of elementary chemical knowledge needed for attending other courses where chemical prerequisites are required. Recognition of elements and compounds in the environment and their impact. Knowledge and prediction of the influence of synthetic compounds on the environment.

Metode poučevanja in učenja:

Learning and teaching methods:

- Predavanja
- Laboratorijske vaje

- Lectures
- Laboratory work

Delež (v %) /

Načini ocenjevanja:

Weight (in %)

Assessment:

<ul style="list-style-type: none"> • Ocena laboratorijskih vaj • Pistni izpit • Ustni izpit 	30	<ul style="list-style-type: none"> • Evaluation of laboratory course • Written exam • Oral exam
	40	
	30	

Reference nosilca / Lecturer's references:

1. DOJER, Brina, PEVEC, Andrej, JAGODIČ, Marko, KRISTL, Matjaž, DROFENIK, Mihael. Three new cobalt(II) carboxylates with 2-, 3- and 4-aminopyridine : syntheses, structures and magnetic properties. *Inorg. Chim. Acta.* [Print ed.], 2012, vol. 383, str. 98-104, doi: [10.1016/j.ica.2011.10.056](https://doi.org/10.1016/j.ica.2011.10.056). [COBISS.SI-ID 15502614]
2. BAN, Irena, KRISTL, Matjaž, DANČ, Valerija, DANČ, Anita, DROFENIK, Mihael. Preparation of cadmium telluride nanoparticles from aqueous solutions by sonochemical method. *Mater. lett.* [Print ed.], 15. Jan. 2012, vol. 67, iss. 1, str. 56-59, doi: [10.1016/j.matlet.2011.09.001](https://doi.org/10.1016/j.matlet.2011.09.001). [COBISS.SI-ID 15371798]
3. KRISTL, Matjaž, GOLOBIČ, Amalija, DOJER, Brina, DROFENIK, Mihael. Synthesis and structure of hydroxylammonium fluoroaluminate. *Monatsh. Chem.*, 2011, vol. 142, no. 8, str. 755-762, doi: [10.1007/s00706-011-0508-4](https://doi.org/10.1007/s00706-011-0508-4). [COBISS.SI-ID 15004182]
4. KRISTL, Matjaž, DOJER, Brina, KASUNIČ, Marta, GOLOBIČ, Amalija, JAGLIČIĆ, Zvonko, DROFENIK, Mihael. Hydroxylammonium fluorometalates : synthesis and characterisation of a new fluorocuprate and fluorocobaltate. *J. fluorine chem.* [Print ed.], Sep. 2010, vol. 131, iss. 9, str. 907-914, doi: [10.1016/j.jfluchem.2010.06.004](https://doi.org/10.1016/j.jfluchem.2010.06.004). [COBISS.SI-ID 14192662]
5. KRISTL, Matjaž, BAN, Irena, DANČ, Anita, DANČ, Valerija, DROFENIK, Mihael. A sonochemical method for the preparation of cadmium sulfide and cadmium selenide nanoparticles in aqueous solutions. *Ultrason. sonochem.* [Print ed.], June 2010, vol. 17, iss. 5, str. 916-922, doi: [10.1016/j.ultsonch.2009.12.013](https://doi.org/10.1016/j.ultsonch.2009.12.013). [COBISS.SI-ID 13766422]